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since the solution contains dilute hcl(aq) and naoh(aq), ... copyright © 2012 nelson education ltd. pre-publication material chapter 5: thermochemistry 5.2-9

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### **CHAPTER 7 CHEMICAL SYSTEMS IN EQUILIBRIUM**

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chapter 7 chemicals systems in equilibrium reflect on your learning (page 422) 1. this is not an example of dynamic equilibrium because there are no opposing processes ...

### **7.3 QUALITATIVE CHANGES IN EQUILIBRIUM SYSTEMS**

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evaluation (c) the addition of hydrochloric acid also increased the total volume of the solution, decreasing the concentration of entities in the equilibrium.

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